## Plasma Materials Interaction Issues For Burning Plasma Experiments

#### M. Ulrickson Presented to the National Research Council Solid State Sciences Committee

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### Outline

- Introduction to Burning Plasmas
- Plasma Materials Interaction Phenomena
- Materials Issues
- Summary

### **Benefits of a Burning Plasma Experiment**

- Study the physics of self heated plasmas
  - Alpha particle heating
  - Alpha energy driven MHD activity
  - Self organized current distribution
- Demonstrate ignition and burn in a magnetic confinement configuration
- Establish the practical components needed for energy production
  - Fueling and particle control
  - Steady-state heat removal
  - Tritium breeding
  - Resistance to neutron damage

### The FIRE Burning Plasma Device

- A compact high field and density tokamak machine
- Major radius 2 m
- Minor radius 0.5 m
- Elongation 1.8
- Magnetic field 8 Tesla
- Density 10<sup>21</sup>/m<sup>3</sup>
- 200 MW fusion power
- 18 s pulse length
- 3000 full power pulses



### The International Thermonuclear Experimental Reactor (ITER)



- Joint design by US, Europe, Japan, and Russia (US dropped out in 1998)
- Superconducting magnets
- 500-1000 MW fusion power
- Fusion gain of 10
- Maximum pulse length 1000 s
- Actively cooled internal components
- Designed for full remote maintenance

# Plasma Materials Interaction Phenomena

### **Fusion Plasma Materials Interactions**

- The core plasma must be kept clean of impurities and He ash
- The plasma facing component surface sees high density and temperature plasma
- Key issues are hydrogen trapping, erosion, and thermal fatigue
- Spans science specialties from ionized gases to materials science



### Science Needed for Fusion Plasma Materials Interactions

- Atomic and molecular physics for ionization, dissociation, and photon radiation of plasma and impurity species
- Surface physics for sputtering, chemical erosion, hydrogen trapping and release, surface segregation
- Materials science for nuclear radiation damage, thermal fatigue, stress corrosion, creep, bonding, and hydrogen trapping
- Engineering science for stress management, heat transfer, and component design

### Understanding of Hydrocarbon Molecule Transport in the Plasma Edge



### Comparison of Erosion Modeling and Experiment



•Erosion data from DIII-D divertor probe under the strike point

•Calculation from the WBC code including sputtering, ionization, transport and redeposition

### **VFTRIM-3D (Vectorized Fractal TRIM)**

- A binary-collision approximation with atomicscale surface roughness using a fractal algorithm.
- Uses a binary collision based on the Kr-C interaction potential and classical scattering kinematics.
- Electronic inelastic energy loss model uses an equipartition between the local Oen-Robinson model and non-local Lindhard-Sharff model.



# Carbon



•Hydrogen insoluble in carbon.

 Implanted H quickly forms a saturated layer in the implant zone. H atoms diffuse rapidly along the porosity (low T).

- •At higher temperatures, the atoms enter into the grains where many are trapped.
- Sputtered carbon can join with D/T to form a stable film on surfaces (codeposition)
- Codeposition traps as much as 40% of all D/T

### **Beryllium**



Russian Academy of Sciences

- Hydrogen isotopes are insoluble in beryllium.
- Implanted H comes out of solution to form a "worm-like" structure several microns deep.
- All H subsequently implanted is released.
- T bred in the beryllium due to n reactions is trapped.
- This T stays in the Be until removal from the reactor.
- A T inventory in the hundreds of grams is possible.

### Tungsten



Blisters on W after exposure on the Tritium Plasma Experiment

- Rapid diffusion and release of tritium prevent significant buildup of tritium in tungsten.
- The low solubility of the hydrogen isotope in the tungsten can result in bubble and blister formation for intense plasma exposure.
- Blister formation could result in the deposition of tungsten into the plasma (flaking and melting).

### **Measurement of Tritium**

- Count betas using a PIN diode.
  - Beta energy is low (<18 keV), short range, only detects near-surface T.
  - Sensitive ~  $10^8$  T/cm<sup>2</sup> or 0.1 Bq/cm<sup>2</sup>.
- Elastic Recoil Detection using MeV ions or neutrons.
  - Measures H,D and T
  - Greater range but less sensitive than beta counting.







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### **Liquid Surface Composition**

#### Experiment







#### Liquid Lithium

Oxygen segregates to the surface upon melting

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### Variation of Sn/Li Liquid Surface Composition



- Sn80Li20 liquid
- Composition measured by small angle scattering
- Red is Sn, blue is Li and green is O
- Segregation of Li on the surface is clearly seen above the melting point

### **Plasma Interaction With a Liquid Li Surface**



- PISCES plasma device
- Lithium light from the interaction of the incident plasma with the evaporated or sputtered Li from the liquid surface
- Studies of erosion rates, temperature limits and hydrogen isotope retention in Li have been conducted.

# **Materials Issues**

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### **Magnetic Fusion Energy Heat Fluxes**



### **Stress Minimization Analysis**

#### **ABAQUS Finite Element Model**

Tungsten, 5 mm thick



- 2-D plane stress
- Elastic behavior
- Temp. dependent props.
- 2000 elements (8-node quad)



### **Progress in PFC Capability**



**Progress:** 

- Reduction of stress using rods on the surface
- Low temperature joining
- Improved heat transfer enhancement

### **Analysis of Disruption Heating**



- Heights code package
- Includes
  evaporation
  and plasma
  shielding
  effects
- Experimentally verified

### **Liquid Surfaces for Fusion Devices**

- Eliminates the erosion issue for component lifetime
- No thermal stress issues
- Some liquids offer particle removal capability
- No neutron damage issues
- Complicated MHD effects (3D magnetic fields that are time varying, fast moving conducting liquids, etc.)
- Temperature limits may be low (heat flux limits)

### **Temperature Limits**



### **Allowed Duration for High T Limit**

![](_page_26_Figure_1.jpeg)

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### Particle Pumping by Liquid Li

- The recombination rate for H on Li is very small (10<sup>-27</sup> to 10<sup>-31</sup> cm<sup>4</sup>/s)
- Several experiments have confirmed nearly 100% retention
- An area of flowing lithium has a large capacity for pumping H isotopes
- 1 m<sup>2</sup> Li flowing at 10 m/s can pump up to 10<sup>23</sup> particles/s
- This particle removal capability is attracting the interest of even existing fusion machines.

### Summary

- The plasma materials interaction issues for a burning plasma device are well defined
  - Tritium retention (hydrogen in materials)
  - Erosion of plasma facing materials (sputtering and chemical)
  - Transport of eroded material (atomic and molecular physics)
  - Cyclic thermal stress (materials and engineering science)
- A substantial experimental database exists to calibrate physics models of the important phenomena

### Summary

- Physics and engineering based models of the important phenomena are being developed and compared to the experimental data
- The extrapolation to a burning plasma device is less of a step than was made when designing the last generation of fusion devices.
- Three potential solutions for PFCs exist and research is being conducted to verify those solutions.

### **Potential Solutions for PFCs**

- All metal water cooled solution
  - W rod surface divertor targets and Be first wall
  - Water cooled copper substrates
  - Nearly completely demonstrated on existing devices (lowest risk)
- Helium gas cooled all refractory metal solution
  - Relies on impurity seeding in the divertor to reduce heat loads (may have no erosion)
  - Uses refractory metal improvement from material program
  - High temperature gas turbine cycle

### **Potential Solutions for PFCs**

- Liquid metal plasma facing components
  - Least developed and greatest risk of insurmountable problems
  - Requires solution of very difficult MHD problems
  - No erosion issues
  - Some materials can pump particles
  - No thermal stress issues
  - Robust to transients
  - Long term, high risk, high payoff research